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Dedicated to Prof. Dr. Wolfgang Pfleiderer on the occasion of his 75th birthday

The E. coli BMT-4D/1A cells have been selected according to Munch-Petersen et al. They carry two regulatory mutations (cytR and deoR) and are able to synthesize constitutively nucleoside-catabolizing enzymes, e.g., cells that possess high UPase and PNPase activities. The cells have been cross-linked by glutaraldehyde to afford a biocatalyst that retained high UPase and PNPase activities and was comfortable for repeated use. An incubation of 2'-deoxyguanosine  $(1)$  and 2-chloroadenine  $(2)$  (molar ratio 3:1) in Kphosphate buffer (10 mm; pH 7.0) in the presence of the biocatalyst at  $65^{\circ}$  for 7 h resulted in quantitative transformation of 2 into 2-chloro-2'-deoxyadenosine (4; cladribine) that was isolated in 81% yield (Scheme 1). Similarly, the reaction of guanosine  $(5)$  and 1,2,4-triazole-3-carboxamide  $(6)$  (molar ratio 1:1) in K-phosphate buffer (10 mm; pH 7.0) in the presence of the biocatalyst at 60 $^{\circ}$  for 30 h led to the formation of 1-( $\beta$ -Dribofuranosyl)-1,2,4-triazole-3-carboxamide  $(8; *ribavirin*)$  in 90-92% yield  $(67-70\%$  isolated yield) (Scheme 2).

**Introduction.** – Nucleoside analogues are a mainstream in the treatment of many viral infections and play a significant role in the chemotherapy of different malignancies  $[1 - 4]$ . Most of them are produced by chemical methods employing the natural nucleosides as starting compounds, or by a convergent approach via the condensation of the carbohydrate precursor and heterocyclic base. The latter strategy suffers from many drawbacks, most serious of which are the low stereospecificity of the glycosidic-bond formation and ambiguous regiospecificity, especially in the case of purine bases. As a result, chemical schemes of nucleoside-analogue preparation are rather complicated and accompanied by a tedious isolation of the individual desired compounds [5].

Considerable progress in the preparation of nucleoside analogues under consideration was achieved by combination of chemical methods and biochemical transformations (for reviews, see [6]). The enzymatic equilibrium-transfer reaction of the sugar moiety of one nucleoside to the heterocyclic base was discovered by *Krenitsky* in the  $60\degree$  [7] [8]. It was established that this equilibrium reaction is catalyzed by purine nucleoside phosphorylase (PNPase) in the case of purine nucleosides and purine bases, and by uridine (thymidine) phosphorylases (UPase or TPase) in the case of pyrimidine nucleosides and pyrimidines. More than a decade later, it was shown that this reaction can be used for the preparation of base- and sugar-modified nucleosides  $[9-13]$ .

Employment of enzymes or the whole cells as biocatalysts results in a principal improvement of key stages of the preparation of nucleoside analogues, viz., allowing fulfillment of the stereo- and regioselective formation of the glycosidic bond. From the published enzymatic nucleoside syntheses, it becomes obvious that the application of chemical-enzymatic methodology for the preparation of nucleosides of medicinal importance may essentially contribute to an improvement of cost-effectiveness ratios vs. current chemical processes. At the same time, despite the widespread appreciation of benefits of  $-$ chemical biotechnology', a number of problems should be overcome in order to displace synthetic methods, which have, in some cases, a highly established position. Biocatalysts should possess higher enzymatic efficiency and operational stability, and need to be more cost-effective.

Two approaches have successfully been employed for the pentofuranosyl-transfer reaction catalyzed by nucleoside phosphorylases: one of these makes use of the purified enzymes, the other is based on the use of the whole bacterial cells displaying the UPase and/or TPase, and PNPase activities as a biocatalyst [6]. During the course of our studies directed towards the search of bacterial cells manifesting the high UPase or/and TPase and PNPase activities, we have selected three strains of  $E$ . *coli*. The first of them, E. coli BM-11, displayed a rather broad specificity in relation to the bases and sugarmodified nucleosides [14] and was successfully employed for the synthesis of a variety of base- and sugar-modified nucleosides  $[15 - 19]$ . It was found that the utilization of cells cross-linked by glutaraldehyde (GA) offers some experimental advantages over unmodified E. coli BM-11 cells  $[17-19]$ . Furthermore, the E. coli BM-11 cells display, besides the high UPase or/and TPase and PNPase activities, high cytidine deaminase (CDase) activity that enables the use of cytosine nucleosides as donors of pentofuranose moiety in the enzymatic transglycosylation. These findings have been successfully used in the synthesis of  $9-(\beta$ -D-arabinofuranosyl)guanine (ara-G), employing 1- $(\beta$ -D-arabinofuranosyl)cytosine (ara-C) as a donor of the arabinofuranose residue and guanosine (Guo) or 2-deoxyguanosine (dGuo) as depot forms of guanine via a cascade of enzymatic reactions [18]. The second strain, E. coli BMT-1D/1A, proved to be somewhat more efficient in the synthesis of a) thymidine, using either i) dGuo,  $2'$ deoxycytidine (dCyd), or 2'-deoxyadenosine (dAdo), or  $ii$ ) a mixture of the same 2'deoxynucleosides resulting from an enzymatic hydrolysis of DNA as donors of 2 deoxyribofuranose moiety  $[20]$ , b) dAdo  $[21]$ , c) a panel of ribo- and 2-deoxyribofuranosides of 1-deaza- and 3-deazapurines  $[22]$  and benzimidazoles  $[22][23]$ , and d) 2chloro-2'-deoxyadenosine (cladribine) [24]. Finally, a slightly modified strain, E. coli BMT-4D/1A, was found to be essentially more efficient vs. the parent E, coli BMT-1D/ 1A in the synthesis of 2'-deoxy-2'-fluoroguanosine [25] and 2,6-diamino-9-(3-deoxy- $\beta$ ---erythro-pentofuranosyl)purine [26].

**Results and Discussion.** – The present study was undertaken in order to develop a practical synthesis of 2-chloro-2'-deoxyadenosine (4) and  $1-(\beta$ -D-ribofuranosyl)-1,2,4triazole-3-carboxamide (8; ribavirin), using dGuo or Guo as donors in the enzymatic transglycosylation of 2-chloroadenine (2; 2Cl-Ade) and 1,2,4-triazole-3-carboxamide (6; TCA), respectively. A choice of these analogues is undoubtedly evoked by their medicinal importance (see, e.g., the leading references [27] and [28], resp.). The E. coli BMT-4D/1A cells have been cross-linked by glutaraldehyde (GA-cells) to afford a biocatalyst that retained high UPase and PNPase activities and was suitable for repeated use. The cross-linking of the cells by GA seems to be advantageous over the alginate gel-entrapped cells used by Holy and co-workers [29] for a related biocatalyst because the GA cells retain operational stability and catalytic activity up to  $65^{\circ}$  in phosphate buffer. Moreover, the GA cells have been shown to be more efficient vs. unmodified cells in the synthesis of nucleosides  $[18]$  and may be used up to five  $-$  ten times in reactions performed at  $60-65^{\circ}$  [30].

The efficient chemical method for the synthesis of cladribine consists in the stereospecific glycosylation of sodium salt of 2,6-dichloropurine with 2-deoxy-3,5-di- $O-p$ -toluoyl- $\alpha$ -D-ribofuranosyl chloride leading to the formation of the corresponding  $N^9$ - $\beta$ -D-nucleoside as the main product (58%), along with the  $N^9$ - $\alpha$ -D-isomer (13%) [31] [32]. Treatment of the former with  $NH<sub>3</sub>$  afforded *cladribine* (71%) [31] [32]. Two enzymatic approaches have been used for the preparation of cladribine. First of them consists in the direct transfer of the 2-deoxyribofuranose residue from thymidine to 2Cl-Ade catalyzed by partially purified *trans-N*-deoxyribosylase [33]. The other, which seems to be advantageous over the former, makes use of PNPase and UPase of E. coli cells (Table 1).

In the present study, we have essentially optimized the conditions of the reaction and employed the more efficient biocatalyst GA-E. coli BMT 4D/1A (Table 1). The enzymatic transdeoxyribosylation studied here includes two coupled reversible reactions, which occur in situ: i) the phosphorolysis of the glycosyl donor, dGuo, in the presence of inorganic phosphate, resulting in the formation of guanine (Gua) and the key intermediate 2-deoxy- $\alpha$ -D-ribofuranose-1-O-phosphate (3), and *ii*) the condensation of the latter with 2Cl-Ade with the release of phosphate affording the desired *cladribine* (4; Scheme 1). Both reactions are catalyzed by PNPase. From the experimental viewpoint, the reaction was conducted in K-phosphate buffer  $(10 \text{ mm};pH)$ 



Biocatalyst: glutaraldehyde-treated E. coli BMT 4D/1A cells.





<sup>a</sup>) The 2-chloroadenosine was used as a source of 2Cl-Ade after in situ phosphorolysis by PNPase, followed by the enzymatic reaction of 2Cl-Ade with 2-deoxy- $\alpha$ -D-ribofuranose-1-O-phosphate resulting from the dAdo phosphorolysis, eventually affording cladribine; an excess of dAdo was transformed into 2'-deoxyinosine by the treatment with adenosine deaminase (ADase) for 30 min at 30 $^{\circ}$ . b) The isolated yield of individual *cladribine* after crystallization was not indicated. <sup>c</sup>) 2Cl-Ade was not substrate of PNPase of E. coli SPT<sup>-</sup>, and 2-chloro-N<sup>6</sup>-(dimethylaminomethylene)adenine was, therefore, used as an acceptor of the 2-deoxypentofuranose residue; alginate gel-entrapped cells were used as a biocatalyst.

7.5) at  $65^{\circ}$  and separated into two steps. First, the reaction was continued for 4 h, affording the main quantity of  $4(75-80%)$ , then the reaction mixture was stored at  $+4^{\circ}$  for 16–18 h, the biocatalyst as well as dGuo and 2Cl-Ade were filtered off, whereas the most of 4 remained in the filtrate. The second step was conducted under the same conditions in a fresh buffer with the biocatalyst, dGuo, and 2Cl-Ade filtered off previously, giving rise to the formation of an additional quantity of  $4(15-20\%)$ . Finally, 4 was purified by silica-gel column chromatography to give, after crystallization from EtOH, the desired compound in 81% isolated yield.

Ribavirin was first synthesised in the 1972 by the chemical glycosylation method [36]. The synthesis is multistep and leads to the formation of two regioisomers that have been separated into individual compounds by chromatography. It should be stressed that only one isomer, *ribavirin*, displayed a broad spectrum of antiviral activity, whereas the other regioisomer is devoid of any biological activity. As distinct from the chemical synthesis, biotechnological process consists in the enzymatic regioselective transfer of the ribofuranose moiety of readily available and inexpensive natural ribonucleosides to the heterocyclic base (6, TCA), affording the desired ribavirin  $(8; Scheme 2)$ . Data on the biotechnological methods are collected in Table 2.

As in the case of the *cladribine* (4) synthesis, the enzymatic transribosylation of TCA consisted in two coupled reversible reactions: i) the phosphorolysis of the glycosyl donor, Guo, in the presence of inorganic phosphate, resulting in the formation of Gua and the key intermediate  $\alpha$ -D-ribofuranose-1-O-phosphate (7), and  $ii$ ) the condensation of the latter with TCA with the release of phosphate, affording the desired ribavirine  $(8; \text{Scheme } 2)$ . Again, both reactions were catalyzed by PNPase. However,





Biocatalyst: glutaraldehyde-treated E. coli BMT 4D/1A cells.





an application of GA-E. coli BMT 4D/1A as a biocatalyst led to displacement of the total equilibrium to the formation of  $8(90-92%)$ . Purification of  $8$  was performed by double ion-exchange chromatography (first *Dowex*  $50 \times 8$ , H<sup>+</sup> form, and then *Dowex*  $1 \times 8$ , OH<sup>-</sup> form) to give, after crystallization from EtOH, the desired compound in  $67 - 70\%$  isolated yield. Structure of the synthesized compound 8 was established by comparison of the m.p., UV, and <sup>1</sup> H-NMR data with those published in [36].

Conclusions. - We have developed a very efficient biocatalyst for the preparation of modified nucleosides. The biocatalyst represents the glutaraldehyde-cross-linked cells of E. coli BMT 4D/1A strain containing very active PNPase, UPase, and TPase. The synthetic potential of the biocatalyst is manifested by the practical preparation of two medicinal drugs *cladribine* and *ribavirin*. Taking into account that the biocatalyst was earlier successfully used for the preparation of 2-deoxy-2-fluoroguanosine [25] and 2,6-diamino-9-(3-deoxy-*β*-*D-erythro-pentofuranosyl)purine* [26], it may be considered as the universal efficient tool for the preparation of modified nucleosides, employing sugar-modified pyrimidine nucleosides as donors of the pentofuranose residue.

## Experimental Part

1. General. High-performance liquid chromatography (HPLC) was carried out with the Waters apparatus (Waters, USA) (column Nova-Pac C-18 (3.9 × 300 mm, Waters); Waters Pump Control Module, Waters 515 HPLC pump; Waters 996 photodiode array detector operating at 208 and 254 nm for analysis of ribavirin and *cladribine*, resp.) and an isocratic elution with 3% MeCN (*ribavirin*) and 7% MeCN (*cladribine*) in H<sub>2</sub>O ( $v/v$ ) at a flow rate 0.7 ml/min (time of analysis was 15 and 30 min, resp.). Guanosine and 2-deoxyguanosine were purchased from Fluka (Switzerland). Column chromatography (CC): silica gel 60 H (70 - 230 mesh ASTM; Merck, Germany). TLC: Silufol UV<sub>254</sub> aluminum sheets silica gel 60  $F_{254}$  (Merck, Germany). Solvent systems for TLC: CHCl<sub>3</sub>/EtOH 4:1 (v/v) (A), BuOH/25% aq. NH<sub>2</sub> 7:2 (v/v) (B). M.p.: *Boetius* apparatus (Germany); not corrected. UV Spectra were measured with Specord M-400 spectrometer (Carl Zeiss, Germany). <sup>1</sup>H- and <sup>13</sup>C-NMR spectra were measured at 200.13 and 50.325 MHz, resp., at 23 $^{\circ}$  on an AC-200 spectrometer equipped with an Aspect 3000 data system (Bruker, Germany);  $\delta$  values are in ppm downfield from internal Me<sub>4</sub>Si (<sup>1</sup>H, <sup>13</sup>C); coupling constants *J* are given in Hz; assignments of proton resonances were confirmed, when possible, by selective homonuclear decoupling experiments.

2. Microorganisms. The whole E. coli BMT-4D/1A cells were used as the source of PNPase. The E. coli BMT-4D/1A cells have been selected according to the technique offered by Munch-Petersen et al. [41]. They carry two regulatory mutations (cytR and deoR) and are able to synthesize constitutively nucleosidecatabolizing enzymes. The biomass of E. coli BMT-4D/1A cells possessing high PNPase activities was prepared as described in [18].

3. Glutaraldehyde Treatment. Cell suspension (4% dry weight) in 0.1 phosphate buffer (pH 7.0) was cooled to 4°. A 25% aq. soln. of GA was added with stirring to give a final concentration of 0.75%. Afterwards, the suspension was incubated for 15 min at the same temp. Then the cross-linked cells were washed free of GA by centrifugation five times with 30 volumes of 10 mm phosphate buffer (pH 7.0).

4. Synthesis of 2-Chloro-2'-deoxyadenosine (= Cladribine; 4). To the heterogeneous mixture (250 ml; 10 m K-phosphate buffer, pH 7.5) consisting of dGuo (1; 8.0 g, 0.03 mol) and 2Cl-Ade (2) [42] (1.7 g, 0.01 mol) at 65°, the biocatalyst (final concentration 0.08% that corresponds to *ca*. 200 mg of the cells calculating for dry weight) was added, and the reaction mixture was incubated at  $65^{\circ}$  for 4 h with gentle stirring. The formation of the product was monitored by TLC on Silufol  $UV_{254}$  plates (A; double elution): the  $R_f$  values for dGuo, 2Cl-Ade, and 4 are 0.13, 0.33, and 0.23, resp. The products were eluted from the TLC plates with H<sub>2</sub>O and quantified spectrophotometrically; under indicated conditions, the formation of  $4$  attained  $75 - 80\%$ . The reaction mixture was stored at  $+4^{\circ}$  for 16–18 h, the formed precipitate, which contained the biocatalyst, as well as dGuo and 2Cl-Ade according to TLC, was filtered off and washed with cold  $(+4^{\circ})$  H<sub>2</sub>O (2 × 40 ml). The filtrate and washings were combined (first H<sub>2</sub>O phase), the residue was suspended in K-phosphate buffer (50 ml; 10.0 mm, pH 7.5), and the mixture was incubated again at 65° for 3 h with gentle stirring. The formation of an additional quantity of  $4(15-20\%)$ ; determined spectrophotometrically) was monitored.

The biocatalyst was removed from the reaction mixture by centrifugation (8000  $\times$  g, 5 min). The precipitate was suspended in H<sub>2</sub>O (40 ml) with stirring for 15 min, and the biocatalyst was removed as described above. Two supernatants were combined with the first H<sub>2</sub>O phase, evaporated to dryness, and the crude 4 was purified by silica-gel CC as described in [23]. The fractions containing individual 4 were combined and evaporated to dryness, the residue was crystallized from EtOH to afford 4 (2.3 g, 81% based on 2Cl-Ade taken into reaction) as a white powder that was identical with the earlier synthesized compound [23] in all respects (TLC, UV, and <sup>1</sup>H-NMR). HPLC:  $t_R$  15.39 min (purity 99.60%).

5.  $1-(\beta$ -D-ribofuranosyl)-1,2,4-triazole-3-carboxamide (8). The reaction mixture (100 ml; 10 mm K-phosphate buffer, pH 7.0) consisting of the biocatalyst (final concentration 1.5% that corresponds ca. 15 mg/ml of the cells calculating for dry weight), Guo (5; 8.5 g, 0.03 mol), and TCA (6) [43] (3.36 g, 0.03 mol)

was incubated at 60° for 30 h with gentle stirring. The formation of the product was monitored by TLC on Silufol  $UV_{254}$  plates (B): the R<sub>f</sub> values for TCA, **8**, and Guo are 0.22, 0.15, and 0.11, resp. The products were quantified spectrophotometrically, and, under indicated conditions, the formation of the desired compound  $\bf{8}$  attained 90  $-$ 92%. The cells were removed from the reaction mixture by centrifugation (8000  $\times$  g, 10 min), washed with H<sub>2</sub>O  $(2 \times 40 \text{ ml})$  with the subsequent withdrawal of the cells by centrifugation, and the combined supernatants were loaded onto a column (4.0  $\times$  20.0 cm; *Dowex* 50  $\times$  8, 200 – 400 mesh; H<sup>+</sup> form). The column was washed with H<sub>2</sub>O, slightly acidic (pH 3.0 - 4.0) *ribavirin*-containing fractions (TLC control) were combined (ca. 300 ml), and loaded onto a column  $(2.5 \times 20.0 \text{ cm})$ ;  $Dowex 1 \times 8$ , 200–400 mesh; OH<sup>-</sup> form). The column was washed with H<sub>2</sub>O, the *ribavirin*-containing fractions (TLC control) were combined (ca. 1.01), and evaporated to dryness. The residue was dissolved under reflux in EtOH (70-80 ml), filtered through a glass filter, allowed to cool to r.t. and left in a refrigerator overnight. Crystalline product was filtered off, washed with cold EtOH, and dried at 60 $^{\circ}$ under vacuum to give  $8(4.91 - 5.13 \text{ g}, 67 - 70\%; \text{TLC-} \text{ and HPLC pure})$ . M.p. 176-178°. UV  $(\text{H}_2\text{O})$ :  $\lambda_{\text{max}}$  207  $(11600)$  ([36]: m.p. 166 – 168° (from aq. EtOH); 174 – 176° (from EtOH)). HPLC:  $t<sub>R</sub>$  3.99 min (purity 99.46%).  ${}^{1}$ H-NMR ((D<sub>6</sub>)DMSO): 8.90 (s, H – C(5)), 7.90 (br. s, 1 H, 1 H of the CONH<sub>2</sub> group presumably H-bonded with N(2)); 7.64 (br. s, 1 H, non-H-bonded H of the CONH<sub>2</sub> group); 5.82 (d,  $J(1',2') = 3.9$ , H-C(1')); 5.62 (d,  $J(H-C(2', OH-C(2'))=5.5, OH-C(2'))$ ; 5.24 (br. s,  $J(H-C(3'), OH-C(3'))=5.5, OH-C(3'))$ ; 4.98 (t,  $J(H-C(5'), OH-C(5')) = 5.5, OH-C(5'))$ ; 4.38  $(m, J(2',3'') = 4.6, H-C(2'))$ ; 4.18  $(m, J(3',4'') = 4.8, H-C(3'))$ ;  $3.98 \ (m, J(4',5') = J(4',5'') = 5.0, H - C(4'))$ ;  $3.60 \ (m, J(5',5'') = 12, H - C(5'))$ ;  $3.48 \ (m, H - C(5''))$ .

## **REFERENCES**

- [1] E. De Clercq, Nature Rev. 2002, 1, 13, and refs. cit. therein; E. De Clercq, Int. J. Antimicrob. Agents 2001, 18, 309; E. De Clercq, J. Clin. Virol. 2001, 22, 73.
- [2] E. De Clercq, Current Med. Chem. 2001, 8, 1543; E. De Clercq, Clin. Microbiol. Rev. 2001, 14, 382; E. De Clercq, Nucleosides Nucleotides Nucleic Acids 2000, 19, 1531.
- [3] J. A. Montgomery, Publ. Am. Inst. Hist. Pharm. 1997, 16, 185.
- [4] M. A. Weiss, Current Oncol. Rep. 2001, 3, 217; V. Gandhi, W. Plunkett, S. Weller, M. Du, M. Ayres, C. O. Rodriguez Jr., P. Ramakrishna, G. L. Rosner, J. P. Hodge, S. O'Brien, M. J. Keating, J. Clin. Oncol. 2001, 19, 2142
- [5] H. Vorbruggen, C. Ruh-Pohlenz,  $\epsilon$  Handbook of Nucleoside Synthesis', John Wiley & Sons, Inc., 2001, pp 646.
- [6] J. R. Hanrahan, D. W. Hutchinson, J Biotechnol. 1992, 23, 193; D. W. Hutchinson, Trends Biotechnol. 1990, 8, 348; T. Utagawa, J. Mol. Catalysis B: Enzymatic 1999, 6, 215.
- [7] T. A. Krenitsky, Mol. Pharmacol. 1967, 3, 526.
- [8] T. A. Krenitsky, J. Biol. Chem. 1968, 243, 2871.
- [9] T. Utagawa, H. Morisawa, T. Nakamatsu, A. Yamazaki, S. Yamanaka, FEBS Lett. 1980, 119, 101.
- [10] T. Utagawa, H. Morisawa, T. T. Miyoshi, F. Yoshinaga, A. Yamazaki, K. Mitsugi, FEBS Lett. 1980, 109, 261.
- [11] T. A. Krenitsky, G. W. Koszalka, J. V. Tuttle, *Biochemistry* 1981, 20, 3615.
- [12] J. L. Rideout, T. A. Krenitsky, G. W. Koszalka, N. K. Cohn, E. Y. Chao, G. B. Elion, V. S. Latter, R. B. Williams, J. Med. Chem. 1982, 25, 1040.
- [13] T. A. Krenitsky, G. A. Freeman, S. R. Shaver, L. M. Beacham 3rd, S. Hurlbert, N. K. Cohn, L. P. Elwell, J. W. Selway, J. Med. Chem. 1983, 26, 891.
- [14] A. I. Zinchenko, L. A. Eroshevskaya, V. N. Barai, I. A. Mikhailopulo, Nucleic Acids Res. Symp. Ser. No. 18, 1987, 137.
- [15] I. A. Mikhailopulo, E. I. Kvasyuk, V. I. Lyakhovets, I. L. Popov, L. A. Eroshevskaya, V. N. Barai, A. I. Zinchenko, Nucliec Acids Res. Symp. Ser. No. 14, 1984, 291.
- [16] L. A. Eroshevskaya, V. N. Barai, A. I. Zinchenko, E. I. Kvasyuk, I.A. Mikhailopulo, Antibiot. Med. Biotechnol. 1986, 31, 174; Chem. Abstr. 1986, 104, 166850t.
- [17] E. N. Kalinichenko, V. N. Barai, S. B. Bokut, V. V. Romanova, A. I. Zinchenko, G. Herrmann, I. A. Mikhailopulo, Biotechnol. Lett. 1989, 11, 621.
- [18] A. I. Zinchenko, V. N. Barai, S. B. Bokut, E. I. Kvasyuk, I. A. Mikhailopulo, Appl. Microbiol. Biotechnol. 1990, 32, 658.
- [19] G. V. Zaitseva, E. I. Kvasyuk, E. V. Vaaks, V. N. Barai, S. B. Bokut, A. I. Zinchenko, I. A. Mikhailopulo, Nucleosides Nucleotides 1994, 13, 819.
- [20] A. I. Zinchenko, V. N. Barai, S. B. Bokut, N. V. Dudchik, Yu. V. Belyaeva, I. A. Mikhailopulo, Biotechnol. Lett. 1990, 12, 341.
- [21] A. I. Zinchenko, V. N. Barai, S. B. Bokut, N. V. Dudchik, E. I. Kvasyuk, I. A. Mikhailopulo, Biotechnol. Lett. 1991, 13, 87.
- [22] I. A. Mikhailopulo, A. I. Zinchenko, S. B. Bokut, N. V. Dudchik, V. N. Barai, E. N. Kalinichenko, H. Rosemeyer, F. Seela, Biotechnol. Lett. 1992, 14, 885.
- [23] I. A. Mikhailopulo, Z. Kazimierczuk, A. I. Zinchenko, V. N. Barai, V. V. Romanova, L. A. Eroshevskya, Nucleosides Nucleotides 1995, 14, 477.
- [24] I. A. Mikhailopulo, A. I. Zinchenko, Z. Kazimierczuk, V. N. Barai, S. B. Bokut, E. N. Kalinichenko, Nucleosides Nucleotides 1993, 12, 417.
- [25] A. I. Zinchenko, V. N. Barai, N. I. Pavlova, E. I. Boreko, I. A. Mikhailopulo, Nucleosides Nucleotides 1999, 18, 687.
- [26] V. N. Barai, A. I. Zinchenko, L. A. Eroshevskaya, E. V. Zhernosek, E. De Clercq, I. A. Mikhailopulo, Helv. Chim. Acta 2002, 85, 1893.
- [27] K. R. Rai, Semin. Oncol. 1998, 3, 19; C. Tortorella, M. Rovaris, M. Filippi, Current Opin. Investig. Drugs 2001, 2, 1751.
- [28] S. S. Lee, M. Sherman, J. Viral Hepat. 2001, 8, 202; S. P. Lawrence, Adv. Intern. Med. 2000, 45, 65.
- [29] I. Votruba, A. Holy, H. Dvorakova, J. Günter, D. Hockova, H. Hrebabecky, T. Cihlar, M. Masojidkova, Coll. Czech. Chem. Comm. 1994, 59, 2303.
- [30] A. I. Zinchenko, L. A. Eroshevskaya, V. N. Barai, Biotechnol. (Moscow) 1990, 5, 36.
- [31] Z. Kazimierczuk, H. B. Cottam, G. R. Revancar, R. K. Robins, J. Am. Chem. Soc. 1984, 106, 6379.
- [32] G. E. Wright, C. Hildebrand, S. Freese, L. W. Dudicz, Z. Kazimierczuk, J. Org. Chem. 1987, 52, 4617.
- [33] M.-C. Huang, K. Hatfield, A. W. Roetker, J. A. Montgomery, R. L. Blakley, Biochem. Pharmacol. 1981, 30, 2663; D. A. Carson, D. B. Wasson, R. Taetle, A. Yu, Blood 1983, 62, 737.
- [34] W. A. Blank, K. J. Elder, W. P. Gati, A. R. P. Paterson, M. A. Pickard, J. S. Wilson, Biotechnol. Lett. 1992, 14, 669.
- [35] I. A. Mikhailopulo, A. I. Zinchenko, Z. Kazimierczuk, V. N. Barai, S. B. Bokut, E. N. Kalinichenko, Nucleosides Nucleotides 1993, 12, 417.
- [36] J. T. Witkowski, R. K. Robins, R. W. Sidwell, L. N. Simon, J. Med. Chem. 1972, 15, 1150.
- [37] H. Shirae, K. Yokozeki, K. Kubota, Agric. Biol. Chem. 1988, 52, 1233.
- [38] H. Shirae, K. Yokozeki, K. Kubota, Agric. Biol. Chem. 1988, 52, 1499.
- [39] H. Shirae, K. Yokozeki, K. Kubota, Agric. Biol. Chem. 1988, 52, 1777.
- [40] W. J. Hennen, C.-H. Wong, J. Org. Chem. 1989, 54, 4692.
- [41] A. Munch-Petersen, P. Nygaard, K. Hammer-Jespersen, N. Fill, Eur. J. Biochem. 1972, 27, 208.
- [42] G. B. Brown, V. S. Weliky, J. Org. Chem. 1958, 23, 125; G. Elion, G. Hitchings, J. Am. Chem. Soc. 1956, 78, 3508.
- [43] G. I. Cipens, V. Grinsteins, Latvijas PSR Zinatnu Akad. Vestis, Kim. Ser. 1965, 2, 204; Chem. Abstr. 1965, 63, 13243f.; G. Cipens, V. Grinsteins, Khim. Geoterotsikl. Soedin., Akad. Nauk Latv. SSR 1965, 4, 624; Chem. Abstr. 1966, 64, 5074f.

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